

## Mars Atmospheric Chemistry and Methane Measurements

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[Contributions from SAM team (PI Paul Mahaffy) including Sushil Atreya]

KISS Study on

#### **Methane On Mars**

December 7<sup>th</sup> - 11<sup>th</sup> 2015

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### **Mars Atmosphere Today**

- Typical surface pressures 6-8 mbar, scale height 11 km, surface temperatures ~210 K
- Even when above freezing temperature, surface pressure too low for liquid water to form.
- Mainly CO<sub>2</sub> (96%), argon-40 (2%), nitrogen (2%), oxygen (0.15%) and CO (0.06%) Viking, SAM
- Each pole in continuous darkness during winter, when up to 25% of CO<sub>2</sub> condenses at caps, subliming back into atmosphere in spring to produce seasonal cycles in pressure and composition.
- Local dust clouds, global dust storms every 2 years, cirrus, frost- MRO
- Thin ozone layer above Mars southern pole in winter
- Trace gases detected: H<sub>2</sub>O, O<sub>3</sub>, CH<sub>4</sub>, H<sub>2</sub>O<sub>2</sub>,





- Villanueva et al. *Icarus*, 2012
- Atreya, Wong, Catling
- Yung et al.



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• Sci-News.com

HOME ASTRONOMY SPACE EXPLORATION ARCHAEOLOGY PALEONTOLOGY BIOLOGY PH'

#### Scientists Discover Third Ozone Layer in Atmosphere of Mars



« PREVIOL

- Known ozone layers near surface and at 40-60 km
- Montmessin and Lefevre (Nat. Geosciences 2013) use SPICAM-MEX data to discover 3<sup>rd</sup> ozone
- layer at 30-70 km over southern winter pole
- Recombination of O atoms from CO2 photolysis, then transport

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### Mars Atmospheric Loss

- McElroy 1972; McElroy + Nier + Yung 1976
- Viking 1976 nitrogen isotope enrichment provided evidence of atmospheric loss
- Toby Owen D/H and hydrogen escape:

Science 24 June 1988: 1767.

### Deuterium on Mars: The Abundance of HDO and the Value of D/H

Tobias Owen, Jean Pierre Maillard, Catherine de Bergh, Barry L. Lutz

Deuterium on Mars has been detected by the resolution of several Doppler-shifted lines of HDO near 3.7 micrometers in the planet's spectrum. The ratio of deuterium to hydrogen is  $(9 \pm 4) \times 10^{-4}$ ; the abundance of H<sub>2</sub>O was derived from lines near 1.1 micrometers. This ratio is enriched on Mars over the telluric value by a factor of  $6 \pm 3$ . The enrichment implies that hydrogen escaped more rapidly from Mars in the past than it does now, consistent with a dense and warm ancient atmosphere on the planet.

- Mars once had a thicker atmosphere and surface liquid water, lost through:
  - Planet's core cooled and solidified (70% density of Earth), losing magnetic field ~4 Gya
  - Catastrophic collision with large body arrested dynamo effect?
  - Gradual erosion by
    - Jeans KE escape
    - Photochemical production of ions that join with e- to reach escape velocities
    - Solar wind pickup and acceleration of ions along solar wind magnetic field, some returning to atmosphere to energize heavier neutrals to escape in sputtering process.

## Congoing Processes in the Upper Atmosphere MAVEN 2013:

- Upper atmosphere, ionosphere, magnetosphere
- Response to solar and solar-wind events
- Ability of atmospheric molecules and atoms to escape to space



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March 2015 interplanetary coronal mass ejection impact -> Large enhancement in escape rate of ions to space. Jakosky et al. *Science* <u>350</u>, Nov 2015.



#### SAM Atmospheric Isotope Ratios -synergy of mass spectrometry *vs* spectroscopy

	A suite of 8 isotope ratios show early atmospheric escape			
lsotopes		Mars value ‰	Instrument	Reference
δ <sup>38</sup>	Ar <sub>Sun</sub>	310 ± 31	QMS	Atreya et al. (2013, GRL)
δ40	Ar <sub>Earth</sub>	5,419 ± 1013	QMS	Mahaffy et al. (2013, Science)
δ15	$N_{Earth}$	572 ± 82	QMS	Wong et al. (2013, GRL)
δ <sup>13</sup>	C <sub>VPDB</sub>	45 ± 12	QMS	Mahaffy et al. (2013, Science)
δ <sup>13</sup>	C <sub>VPDB</sub>	46 ± 4	TLS	Webster et al. (2013, Science)
δ18	O <sub>SMOW</sub>	48 ± 5	u	и и
δ17	O <sub>SMOW</sub>	24 ± 5	u	и и
δ13	Cδ <sup>18</sup> O	109 ± 31	u	и и
δD	SMOW	4,950 ± 1080	"	и и
u		4,231 ± 33	u	Leshin et al. (2013, Science; updated)



### **Atmosphere-surface Interactions**



- Current day atmospheric d<sup>13</sup>C in CO<sub>2</sub> (+46 per mil) is the result of the history of atmospheric loss (<sup>13</sup>C enrichment) and carbonate deposition (<sup>13</sup>C depletion)
- Hu et al. show that escape of C via CO photodissociation and sputtering enriches <sup>13</sup>C, a process partially compensated by moderate carbonate precipitation.

#### \* Keck Martian Atmosphere Shows Significant Early Loss





#### D/H ratio in Yellowknife Bay Mudstone -Mahaffy et al. Science Dec 2014



- D/H from both evolved water (TLS) and hydrogen (QMS) at 3 x SMOW.
- Low D/H in bound OH in clays formed 2.9-3.5 Gya (Hesperian) shows that considerable water was lost both before and after this point.
- GEL at Cumberland mudstone formation ~150m compared to ~50m today.

![](_page_9_Figure_6.jpeg)

![](_page_10_Picture_0.jpeg)

## Earth-based Observations- H<sub>2</sub>O and HDO

![](_page_10_Figure_2.jpeg)

Villanueva et al., Science 348, 218-221, 2014

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![](_page_11_Picture_0.jpeg)

## Earth-based Observations – $H_2O_2$

Therese Encrenaz (Paris Observatory) et al., *Icarus*, 2004

- Discovered 40 ppbv H<sub>2</sub>O<sub>2</sub> on Mars at subsolar point (white dot) in summer time using the 3 m NASA IRTF at Hawaii
- Spatial and seasonal variations

![](_page_11_Figure_5.jpeg)

### Martian dust devils, electrochemistry and oxidants

Sushil Atreya et al. (2006) – Large turboelectric fields in dust storms can produce greatly enhanced OH and therefore  $H_2O_2$ . Resulting superoxides and other oxidants could destroy methane gas and form formaldehyde or methanol.

![](_page_11_Picture_8.jpeg)

![](_page_12_Figure_0.jpeg)

![](_page_13_Picture_0.jpeg)

## **Spectral Resolution**

![](_page_13_Figure_2.jpeg)

![](_page_14_Picture_0.jpeg)

## Earth-based Observations – CH<sub>4</sub>?

- Aug 1969 One week after Moon landing- in JPL's Von Karman auditorium.... Walter Cronkite
- Mariners 6 and 7 IRS (PI George Pimental) announced discovery of significant NH<sub>3</sub> and CH<sub>4</sub> on Mars.....later retracted....

![](_page_14_Picture_4.jpeg)

![](_page_14_Figure_5.jpeg)

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![](_page_15_Figure_0.jpeg)

## Vittorio Formisano et al.

#### (Institute of Physics & Interplanetary Space)

Geminale et al. PSS 59 (2011) 137-148

- Planetary Fourier Spectrometer on ESA's Mars Express
- Designed to measure vertical profiles of  $CO_2$  and column H<sub>2</sub>O, CO, CH<sub>4</sub> and H<sub>2</sub>CO
- Detected  $15 \pm 5$  ppbv CH<sub>4</sub> in the Martian atmosphere by averaging 15,687 spectra
- CH<sub>4</sub> signatures in 9 lines (!)
- CH<sub>4</sub> mixing ratio variable from 0 to 40 ppbv
- CH<sub>4</sub> and H<sub>2</sub>O emanate from 3 locations: Arabia Terra, Elysium Planum, and Arcadia-Memnomia
- Conclude source is northern cap, whose summertime release is sufficient to explain global mean of 15 ± 5 ppbv.

#### Michael J. Mumma et al. (NASA GSFC) -2004

- Observations mainly from the 8 m Gemini South telescope in Chile (& NASA's IRTF, Mauna Kea)
- Relies on Doppler shifts
- Multiple spectral features of CH<sub>4</sub> seen
  - Need to subtract Earth CH<sub>4</sub>
- Lower amounts at the higher latitudes 20-60 ppbv
- Significant enhancement at equatorial regions up to 40 ppbv – in region of changing topography
   transition highlands to plain, scarps, cliff faces
- Enhancement also over deep rift Valles Marineris

   steep high cliffs
  - Is methane diffusing under the permafrost and emerging at cliff faces or fissures?

![](_page_16_Picture_9.jpeg)

17

![](_page_16_Picture_10.jpeg)

### Vladimir Krasnopolski et al. (Catholic University) et al. *Icarus* 2004, EPSC 2011

![](_page_17_Picture_1.jpeg)

- Detected 10 ± 3 ppbv CH<sub>4</sub> in Mars atmosphere where lifetime is 340 years
- 270 tons CH<sub>4</sub> therefore produced per year
- Concluded that observed CH<sub>4</sub> is from methanogenesis by living subterranean organisms (Martian biota scarce and sterile except oases)

![](_page_17_Figure_5.jpeg)

![](_page_18_Picture_0.jpeg)

- **Conflicting results** from 0 ppbv to 60 ppbv
- Differing distributions uniform distribution to high local variations- and widely differing interpretations

nature	Vol 460 6 August 2009  <b>doi:10.1038/nature08228</b>
LETTERS	

#### **Observed variations of methane on Mars unexplained by known atmospheric chemistry and physics**

Franck Lefèvre<sup>1</sup> & François Forget<sup>2</sup>

So what did TLS-SAM on Curiosity see?

![](_page_18_Picture_7.jpeg)

![](_page_18_Picture_8.jpeg)

# TLS – Tunable Laser Absorption Spectrosco

![](_page_19_Figure_1.jpeg)

![](_page_20_Picture_0.jpeg)

### Tunable Laser Spectrometer (TLS)

![](_page_20_Picture_2.jpeg)

![](_page_20_Picture_3.jpeg)

#### Mass = 3.3 kg

Near-IR TDL from Nanoplus, Germany

IC laser from MDL-JPL

![](_page_20_Picture_7.jpeg)

![](_page_20_Picture_8.jpeg)

![](_page_21_Picture_0.jpeg)

### **TLS-SAM Methane Search**

![](_page_21_Picture_2.jpeg)

- Fore-optics chamber contains residual terrestrial air
- Use "difference method" comparing full sample cell to empty sample cell

![](_page_21_Figure_5.jpeg)

![](_page_21_Picture_6.jpeg)

![](_page_22_Picture_0.jpeg)

## **Direct Ingest Spectra**

![](_page_22_Picture_2.jpeg)

![](_page_22_Figure_3.jpeg)

![](_page_22_Figure_4.jpeg)

- A. "Low methane" from Sols 79, 81, 106, 292, 313 and 684;
- B. "High methane" from Sol 474 only

![](_page_22_Picture_7.jpeg)

![](_page_23_Picture_0.jpeg)

"Mars Methane Detection and Variability at Gale Crater",

Webster et al., *Science*, **347**, 415-417 (2015)

![](_page_23_Figure_1.jpeg)

![](_page_23_Picture_2.jpeg)

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#### **Possible Methane Sources and Sinks** Winds UV 0 C **Cosmic Dust Carbon Dioxide** Photochemistry Methane Surface Organics Outgassing Formaldehyde Methanol Methane Clathrate Subsurface Storage Microbes Methane Olivine (rock) Water

![](_page_25_Picture_0.jpeg)

## Methane Sources to consider

![](_page_25_Picture_2.jpeg)

- Unknown photochemical processes in the atmosphere that may involve dust (heterogeneous chemistry)
- Geological production such as serpentinization of olivine
- UV degradation of IDP's, meteoritically-delivered organics
- Release from gas trapped in subsurface clathrates
- Release from regolith-adsorbed gas
- Erosion of basalt with methane inclusions
- Geothermal production
- Production from sub-surface methanogens

#### Hu.....Yung group (2015):

- Release following deliquescence of perchlorate salts
- Microorganism release of methane from organic decay in solution
- Deep subsurface aquifers that produce bursts of methane as a result of freezing and thawing of the permafrost as in the Arctic – expect seasonal dependence (?)

![](_page_25_Picture_15.jpeg)

![](_page_26_Picture_0.jpeg)

## Methane Sinks

![](_page_26_Picture_2.jpeg)

- Photochemical processes (Krashnopolsky et al., 2004) UV +OH oxidation
- Electrical discharges in dust devils (Farrell et al., 2006)
- Reactions with oxidants in the soil (Atreya et al., 2006).
- Wind erosion of quartz grains Methane is removed by reactions with abraded silicates, which leads to covalent ≡Si-CH<sub>3</sub> bonds and thus an enrichment of the soil with reduced carbon (Jensen et al., 2014; Bak et al. EPL 2015 in press).- could explain the fast disappearance of methane.

![](_page_26_Picture_7.jpeg)

![](_page_27_Picture_0.jpeg)

![](_page_27_Picture_1.jpeg)

## TLS-SAM Low Methane Background 🔤

#### A contribution from a variety of sources?

#### Consider UV degradation of surface material

- Surface meteoric material: micrometeorites from accreted IDP, and carbonaceous chondrites containing few wt% organic matter;
- Accreted IDP's deliver 90% of organic C with average C content of 10%- a carbon-limited (not UV) process.
- UV/CH<sub>4</sub> model of Schuerger et al. (2012) predicts
  - Over geological time, a present day 2.2 ppbv methane for 20% conversion efficiencies of 10 wt% C material
  - Very small diurnal/seasonal changes (daily input is 0.02 pptv)
- TLS-SAM background level is ~3 times < model prediction of ~2.2 ppbv
  - Therefore infall amount, C conversion efficiency or organic content is overestimated by factor of ~3.
  - But lab measurements do NOT accurately simulate Mars conditions, such as inclusion of surface oxidants
  - Steady infall/conversion cannot explain large episodic bursts.

![](_page_27_Picture_14.jpeg)

![](_page_28_Picture_0.jpeg)

![](_page_28_Picture_1.jpeg)

![](_page_28_Picture_2.jpeg)

#### TLS-SAM Methane vs Solar Longitude L<sub>s</sub>

![](_page_28_Figure_4.jpeg)

New mean value = 0.5 ppbv

![](_page_28_Picture_6.jpeg)

![](_page_29_Picture_0.jpeg)

# TLS-SAM Methane vs Solar Longitude L N. Spring N. Summer N. Autumn N. Winter

![](_page_29_Picture_2.jpeg)

![](_page_29_Figure_3.jpeg)

![](_page_29_Picture_4.jpeg)

30

![](_page_30_Picture_0.jpeg)

## **TLS-SAM High Methane**

![](_page_30_Picture_2.jpeg)

- High methane of ~7 ppbv is within range of UV/CH<sub>4</sub> model predictions if transported from region of airburst/bolide, but MRO shows no measurable impacts at Gale Crater since landing.
- No correlations with pressure, surface/air temperature, opacity, UV flux, in situ H<sub>2</sub>O, surface mineralogy/composition.
- Anti-correlations with column water, oxygen now ruled out.
- Consistent with a small local source producing a temporal (not location) change.
- Wind fields and daytime

increase indicate source to the north.

![](_page_30_Picture_9.jpeg)

![](_page_30_Picture_10.jpeg)

![](_page_31_Picture_0.jpeg)

### **Encounters with Cometary Debris Fields?**

![](_page_31_Figure_2.jpeg)

Geochemical Perspectives Letters v2, n1 | doi: 10.7185/geochemlet.1602. Dec 2015.

- Could "meteor showers" deliver macromolecular carbon MMC that produces CH4 under UV photolysis- possibly at higher altitudes?
- Maarten-Roos/ Atreya 2015 conclude no correlation with cometary debris/meteoritic infall.
- Fries et al. (GPL, 2015) use data from PFS, Mumma, MSL to show possible dependence on cometary debris encounters
  - But more recent TLS-SAM measurements show no repetition.

![](_page_32_Picture_0.jpeg)

## Local Source Within Gale Crater?

#### Tracking the MSL-SAM methane detection source location Through Mars Regional Atmospheric Modeling System (MRAMS)

Jorge Pla-García, Scot C.R. Rafkin, and Alberto G. Fairén<sup>2</sup>

Southwest Research Institute, Boulder CO 80302, USA

<sup>2</sup>Centro de Astrobiología (CSIC-INTA) Carretera de Ajalvir, km.4, 28850 Torrejón de Ardoz, Madrid, Spain EGU Abstract 2016

![](_page_32_Figure_6.jpeg)

- Modeling supports local source within Gale Crater with existing chemistry, OR
- Source outside Gale Crater with subsequent deep mixing and rapid destruction

![](_page_33_Picture_0.jpeg)

![](_page_33_Picture_2.jpeg)

"Zahnle, who was also critical of the 2003 and 2004 methane reports, said that it wouldn't take much from the rover to lead scientists astray. After all, the rover contains within a chamber some methane at a concentration 1,000 times higher than the puff supposedly found in Mars' atmosphere. Curiosity's methane comes from Earth" – Discovery News

- TLS foreoptics chamber contains ~12 ppmv methane, or ~2 nanomoles CH<sub>4</sub>, or <u>~10<sup>15</sup> molecules.</u>
- A 10-m diam sphere around the rover with 7 ppbv methane contains 1 micromole of methane, or <u>~10<sup>18</sup> molecules</u>.
- If surface winds are only ~1m/sec, sphere volume is replenished in 10 secs, so to sustain high methane for 2 months, would need a source of ~<u>5 x 10<sup>23</sup> molecules</u>!
- Also, foreoptics chamber CH<sub>4</sub> content shows no evidence of loss over time.

![](_page_33_Picture_8.jpeg)

![](_page_33_Figure_9.jpeg)

![](_page_33_Picture_10.jpeg)

![](_page_34_Picture_0.jpeg)

- The <u>very low background</u> level of methane (~0.7 ppbv) could result from the UV degradation of surface organics, but is 3 times lower than model predictions. A geological or biological source cannot be determined. A seasonal dependence or time decay may be evident.
- The <u>sudden spike</u> in methane (~7 ppbv) indicates a release from either modern production or from storage of older methane in clathrates. Higher daytime values suggest a northerly source. No obvious correlations with oxygen, other species, meteoritic infall, cometary debris. Seasonal repetition not observed.
- The sudden rise in methane and <u>the fact that it came back down quickly</u> indicate the source was most likely relatively localized and small.
- These observations of methane are suggestive of a *currently active Mars*.

![](_page_35_Picture_0.jpeg)

### Advanced TLS Instruments for Mars Methane Isotope Measurements

- TLS-SAM-MSL (Webster, Mahaffy)
  - Sensitivity 0.1 ppbv with enrichment

![](_page_35_Picture_4.jpeg)

- Enhanced pathlength digital TLS (Webster, Flesch et al. JPL)
  - Sensitivity 20 pptv with enrichment
- Cavity Ringdown TLS (Okamura (Caltech), Christensen (JPL))
  - Sensitivity 10 pptv
- Integrated Cavity Output Spectroscopy (ICOS)- Vinogradov (IKI, Russia) for ExoMars 2018 Lander
  - Sensitivity <50 pptv</li>

![](_page_36_Picture_0.jpeg)

- ISRO's first planetary mission after successful Chandrayaan Moon mission
- Entered Mars orbit Sep 2014
- Methane spectrometer will subtract solar reflectance at 1.65  $\mu$ m from that at 3.3  $\mu$ m to produce column CH<sub>4</sub> in a global map.
- Sensitivity undetermined, expected 10 ppbv

![](_page_36_Picture_5.jpeg)

### NOMAD on ExoMars Orbiter

- Nadir and Occultation for MArs Discovery (NOMAD)

   two instruments, echelle grating, 0.2-0.4 cm<sup>-1</sup>
   resolution
- PI is Ann Vandaele, Belgian Institute for Space Aeronomy
- Expected detectivity of ~25 pptv in solar occultation, and 11 ppbv in nadir view.

![](_page_36_Picture_10.jpeg)

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![](_page_37_Picture_0.jpeg)

- Solar occultation orbiting FTIR with 0.03 cm<sup>-1</sup> spectral resolution
- Long pathlength (~10 km), high vertical resolution
- Huge number of spectral lines, IR bands, many molecules simultaneously
- Pl's Paul Wennberg (Caltech) and Vicky Hipkin (Canadian Space Agency), with Drummond, Toon, Allen, Blavier, Brown, Kleinbohl, Abbatt, Lollar, Strong, Walker, Bernath, Clancy, Coutis, DesMarais, Eiler, Yung, Encrenaz, McConnell

![](_page_37_Figure_5.jpeg)

![](_page_37_Figure_6.jpeg)

KISS Study - Methane on Mars, Webster 2015